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Please find below and/or attached an Office communication concerning this application or proceeding.

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**BEFORE THE BOARD OF PATENT APPEALS
AND INTERFERENCES**

Application Number: 10/621,292
Filing Date: July 17, 2003
Appellant(s): SAN ET AL.

MAILED
DEC 18 2007
GROUP 2800

Scott L. Appelbaum
For Appellant

EXAMINER'S ANSWER

This is in response to the appeal brief filed 11/05/2007 appealing from the Office actions mailed 04/03/2007 and 06/13/2007.

(1) Real Party in Interest

A statement identifying by name the real party in interest is contained in the brief.

(2) Related Appeals and Interferences

The examiner is not aware of any related appeals, interferences, or judicial proceedings which will directly affect or be directly affected by or have a bearing on the Board's decision in the pending appeal.

(3) Status of Claims

The statement of the status of claims contained in the brief is correct.

(4) Status of Amendments After Final

The appellant's statement of the status of amendments after final rejection contained in the brief is correct.

(5) Summary of Claimed Subject Matter

The summary of claimed subject matter contained in the brief is correct.

(6) Grounds of Rejection to be Reviewed on Appeal

The appellant's statement of the grounds of rejection to be reviewed on appeal is correct.

(7) Claims Appendix

The copy of the appealed claims contained in the Appendix to the brief is correct.

(8) Evidence Relied Upon

5,196,360	Doan et al.	3-1993
5,766,997	Takeuchi	6-1998

2002/0151170A1	Maex et al.	10-2002
6,503,840 B2	Catabay et al.	1-2003
6,664,166 B1	Jaiswal et al.	12-2003
6,775,046 B2	Hill et al.	8-2004

(9) Grounds of Rejection

The following ground(s) of rejection are applicable to the appealed claims:

Claims 1, 5-6 and 12, 16-17, 27-28 and 31 are rejected under 35 U.S.C. 103(a) as being unpatentable over Doan et al. US 5,196,360 in combination with Takeuchi US 5,766,997 and Maex et al. US Pub. 2002/0151170 A1.

Re claims 1 and 12, the Doan et al. reference discloses a method for fabricating a semiconductor device, figs 1-4, comprising:

forming a field region on a substrate 12 to define an active region; forming a gate pattern 22/14 on the active region, wherein the gate pattern includes sidewalls; forming spacers 24 on the sidewalls of the gate pattern; forming source/drain regions 16/18 aligned with the spacers on both sides of the gate pattern;

*forming a metal film of *titanium* layer 28 for silicide on the entire surface of the substrate;*

*forming an N-rich titanium nitride layer 30 on the *titanium* layer, col. 4, lines 42-54;*

*thermally treating the *titanium* layer 28 for silicide and the N-rich titanium layer 30 to form a *titanium* silicide layer on the gate pattern and the source/ drain region, col. 4, line 55 to col. 5, line 8; and*

selectively removing the *titanium* layer for silicide and the N-rich titanium nitride layer, wherein a top portion of the *titanium* silicide on the gate pattern and the source/drain region is exposed, col. 5, lines 17-21.

The Doan et al. reference uses a *titanium* layer 28 for silicide on the silicon substrate, it does not use Ni-based metal layer comprised of a nickel alloy for silicide on the silicon substrate. The Doan et al. reference also does not disclose cleaning the substrate using a wet cleaning process.

The Takeuchi reference discloses a method for fabricating a semiconductor device, embodiment 4, comprising:

forming a field region on a substrate 121 to define an active region, fig. 12A;
forming a gate pattern 125 on the active region, wherein the gate pattern includes sidewalls, fig. 12B;
forming spacers 130/131 on the sidewalls of the gate pattern, fig. 12D;
forming source/drain regions 127/128, 132/133 aligned with the spacers on both sides of the gate pattern;

"the source region is damaged by ion implantation. Before the silicide layer is formed, therefore, dilute HF cleaning is generally performed to expose the surface of the silicon substrate", col. 9, lines 35-37;

forming nickel or *titanium* or *cobalt* interchangeably, col. 1's lines 29-30, for a metal layer 136 for silicide on the entire surface of the substrate, or a nickel alloy, col. 7's lines 30-37:

a metal which can form silicide when reacted with silicon (this metal will be hereinafter called "silicide forming metal"). This silicide forming metal is, for example, refractory metal, more specifically, at least one kind of metal selected from a group of tungsten (W), cobalt (Co), titanium (Ti) and nickel (Ni). The first metal can be formed by a known thin film forming technology, such as sputtering or CVD.

forming a titanium nitride layer 137 on the Ni-based metal layer 136;

Then, a reaction suppressing layer is formed on the first metal layer including at least above the drain region and excluding above the source region. ...

The reaction suppressing layer is formed of a material which causes no silicidation with silicon; or low-resistance material which may cause a silicidation but has a lower reactivity than the mentioned metal. One example of the material for the reaction suppressing layer is a metal nitride. This metal nitride may be a nitride of the aforementioned silicide forming metal. More specifically, the metal nitride is at least one kind selected from a group of titanium nitride, cobalt nitride, nickel nitride and tungsten nitride. When the reaction suppressing layer is formed of a metal nitride, this metal nitride should not necessarily be a nitride of the same metal as is used for the first metal layer. When a metal nitride is the material for the reaction suppressing layer, this layer may be formed by CVD, sputtering and the other applicable to this process, col. 7, lines 42-62.

thermally treating the Ni-based metal layer *comprised of nickel alloy* for silicide and the titanium nitride layer to form a nickel silicide layer on the gate pattern and the source/drain region, col. 17, lines 24-30; and

selectively removing the Ni-based metal layer for silicide and the titanium nitride layer, wherein a top portion of the nickel silicide on the gate pattern and the source/drain region is exposed, col. 17, lines 39-41.

The Takeuchi reference uses titanium nitride and makes the titanium nitride layer enriched with nitrogen while annealing "under the nitrogen or ammonia environment", col. 8, lines 12-13, it does not directly use N-rich titanium nitride.

It would have been obvious to one of ordinary skill in the art at the time of the invention to provide the titanium layer for silicide of the Doan et al. reference with material of nickel or nickel alloy as taught by Takeuchi because: firstly, the titanium or nickel or nickel alloy layer for silicide are recognized as equivalent materials by Takeuchi; secondly, those materials as stated by Takeuchi would provide the metal layer for silicide of Doan et al. the same characteristic as analyzed by Takeuchi to

enhance the reduction in sheet resistance (Takeuchi's col. 1, lines 24 and 65).

Further, it would have been obvious to one of ordinary skill in the art at the time of the invention to provide the process of Doan et al. with the step of cleaning the substrate using a wet cleaning process as taught by Takeuchi as the cleaning step would mitigate implantation damage on exposed surface of the silicon substrate.

Alternatively, it would have been obvious to one of ordinary skill in the art at the time of the invention to provide the method of Takeuchi with N-rich titanium nitride of Doan because the N-rich titanium nitride of Doan would provide the titanium nitride of Takeuchi with inhibition ability of "outgrowth of silicide and potential short circuit paths between adjacent silicide contact areas" (Doan's abstract).

The combination of Doan et al. and Takeuchi teaches substantially all of the instant steps of fabricating a semiconductor device. The combination does not disclose the nickel alloy layer includes greater than 0 to about 20 % of one of the materials of Ta, Zr, Hf, Pt, Pd, V, Nb or any combination of these (excluding Ti, Co and W, *re claims 27-28 and 31*).

The Maex et al. reference discloses these materials as elements in a Ni alloy used to formed Ni silicide [0014] or [0019], e.g. especially [0014] discloses

The first layer structure can also include a cobalt-nickel alloy with the nickel content varying from 0 to 100%; ... Also, other metals such as Pt or Pd can be chosen as elements that are present in the first layer structure.... or the elements Pt and Pd can be added in minor amounts to the first layer structure. Also, other elements such as Au, Ir, Os, Rh, Ti, Ta, W, Mo, Cr, C, and Ge can be part of the first layer structure.

It would have been obvious to one of ordinary skill in the art at the time of the invention to provide the method of the combination with Maex et al.'s Ni alloy containing

one of Ta, Zr, Hf, Pt, Pd, V, Nb or any combination of these materials, because the Ni alloy of Maex et al. would provide the combination with materials that “can withstand thermal treatment without significant degradation” [0009].

In this combination, choice of ratio of elements in the Ni-based metal layer would have been a matter of routine optimization because elements ratio, amongst many other variable parameters, is known to affect device properties and would depend on the desired device density on the finished wafer and the desired device characteristics. One of ordinary skill in the art would have been led to the recited greater than zero to 20 % of the chosen elements in the alloy through routine experimentation to achieve desired characteristics as suggested by Maex et al.

Further, use of Ni-based metal *comprised of nickel alloy* and N-rich titanium nitride in the combination would provide “the nickel silicide on the gate pattern neither shorted nor cut, a pit is prevented from being formed in a boundary area between the active region and the field region, and lumping of the nickel silicide is prevented and a silicide residue is prevented from remaining on the spacers and the field region” as claimed and well-suited with Doan et al.’s col. 6, line 62 to col. 7, line 6 “for inhibiting outgrowth of adjacent silicide contact areas which have the potential for forming short circuit paths between the silicide contact area” and preventing “pitting of the silicon substrate”, Doan et al.’s col. 4, lines 64-68.

Re claims 5 and 16, in the combination, the Doan et al. reference discloses the chemical formula TiN_x where $x > 1$ or from about 1 to 2 or 1.1 to 1.3 (col. 2, line 8, col. 3, lines 24-31, col. 6, line 8).

Re claims 6 and 17, in the combination, the Takeuchi reference discloses the thermal treatment for forming nickel silicide layer is carried out using a RTN, col. 8, line 11, not in vacuum but obviously must be in a thermal system.

Claims 2, 7-8 and 13, 18-19, 22-23, 26 and 29-30 are rejected under 35 U.S.C. 103(a) as being unpatentable over the Doan et al./Takeuchi/Maex et al. combination as applied to claims 1, 5-6 and 12, 16-17, 27-28 and 31 above, and further in view of Catabay et al. US 6,503,840 B2, Jaiswal et al. US 6,664,166 B1 and Hill et al. US 6,775,046 B2.

The combination of Doan et al./Takeuchi/Maex et al. teaches substantially all of the instant steps of the method for fabricating a semiconductor device. Although Doan et al. teaches the transistor structure is formed using conventional technique, *metal* layer 28 for silicide and nitrogen-rich titanium nitride layer 30 are formed by sputtering (col. 4, lines 2-4 and 35-54), and Takeuchi teaches cleaning the surface of the substrate and forming the Ni-based metal layer comprised of nickel alloy and titanium nitride layer by sputtering; however, none of the reference teaches at what temperature the Ni-based metal layer is formed, and using RF sputtering etching to remove particles from the surface of the substrate in situ with the formation of Ni-based layer and TiN layer.

Re claims 2, 13, 19, 22-23 and 26, the Hill et al. reference teaches

As known, the temperature at which the target is maintained influences the composition of the alloy that is deposited on the substrate during sputtering. As example, if the block of metal in dish 27 is a titanium nickel alloy of 50% titanium and 50% nickel, and that target is at room temperature during the sputtering process, the alloy deposited on the substrate will be different in composition, namely, 48% titanium and 52% nickel. If the target is at 100 degrees C. during the sputtering process, then the composition of the deposited alloy will be 49% titanium and 51%

nickel. And if the target is maintained at a temperature of 200 degrees C. during the sputtering process, the deposited alloy will be 50% titanium and 50% nickel, col. 9, lines 34-45.

Choice of temperature amongst many other variable parameters in the formation of elements would have been a matter of routine optimization because temperature is known to affect process steps and resulting device properties and would depend on the desired device density on the finished wafer and the desired device characteristics as taught by Hill et al. One of ordinary skill in the art would have been led to the recited temperature through routine experimentation to achieve desired deposition and reaction rates. It would have been obvious to one of ordinary skill in the art at the time of the invention to provide the process of the combination with the Ni-based metal *comprised of nickel alloy* sputtering with selected temperature of about 25 to 500 °C in a thermal treatment system because the sputtering of Ni-based metal within the selected temperature range in the system would give the process of the combination with the desired metal as taught by Hill et al.

Re claims 7-8, 18-19, 22-23 and 26, the Catabay et al. reference discloses the process wherein the contaminated surface is solvent cleaned to remove residues and then RF cleaned before titanium and then titanium nitride are deposited over the surface in the same chamber (abstract). And/or the Jaiswal et al. reference discloses "a method for processing a partially fabricated semiconductor wafer ... including performing a wet pre-metallization cleaning step on the surface of the wafer, performing an RF plasma sputter etching process ... while maintaining unbroken vacuum conditions ... and

depositing a layer of metal on the surface of the wafer ... a stabilization bake cycle then is performed on the wafer", col. 2, lines 50-66.

It would have been obvious to one of ordinary skill in the art at the time of the invention to provide the cleaning and depositing of the combination of Doan et al. with Takeuchi and Maex et al. with the teachings of Catabay et al. and/or Jaiswal et al. because the steps of cleaning/etching and depositing of Catabay et al. and/or Jaiswal et al. would provide the process of the combination with continuous process and preventing further contamination.

(10) Response to Argument

Appellant's Arguments in 7.A. (i) states "the Examiner has failed to meet his initial burden of establishing a prima facie showing of obviousness by failing to show that the above combination of cited reference provides sufficient motivation to one skill in the art to provide a method for fabricating a semiconductor device which includes each and every element recited in claims 1, 12, 19, 23 and 31." The examiner does not agree. The test for obviousness is not whether the features of a secondary reference may be bodily incorporated into the structure of the primary reference; nor is it that the claimed invention must be expressly suggested in any one or all of the references. Rather, the test is what the combined teachings of the references would have suggested to those of ordinary skill in the art. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981).

Appellant seems to ignore the argument on the term "alloy" that the examiner provides in the Final Office Action mailed 04/03/2007 and further in the Advisory action

mailed 06/13/2007 with documents support for the arguments attached. That argument is copied herewith from that Advisory:

In response to applicant's statement that "Takeuchi does not expressly mention using alloys for its metal layer for silicide", applicant is directed to the quoted passage on page 3 of the office action mailed 04/03/2007 wherein

a metal which can form silicide when reacted with silicon (this metal will be hereinafter called "silicide forming metal"). This silicide forming metal is, for example, **refractory metal**, more specifically, at least one kind of metal selected from a group of tungsten (W), *cobalt* (Co), *titanium* (Ti) and nickel (Ni). The first metal can be formed by a known thin film forming technology, such as sputtering or CVD, col. 7's lines 30-37.

As defined in Wikipedia of refractory metal mentioned in Takeuchi's passage

Refractory metals are characterized by their extremely high melting points, which range well above those of iron and nickel. When the refractory metals are considered to be those metals melting at temperatures above 2123 K, twelve metals constitute this group: tungsten (the melting point 3683 K), rhenium, osmium, tantalum, molybdenum, iridium, niobium, ruthenium, hafnium, zirconium, vanadium, and chromium.

Applicant is further directed to Class Definitions by Class Number in the Classification used in and out of USPTO office wherein the definition of class 438, subclass 582 is as follow:

Using refractory group metal (i.e., titanium (Ti), zirconium (Zr), hafnium (Hf), vanadium (V), niobium (Nb), tantalum (Ta), chromium (Cr), molybdenum (Mo), tungsten (W), or alloy thereof)

in which all of the materials in the parenthesis belong to refractory group metal. (Both documents are attached in this office action). Therefore, those claimed materials, which are not explicitly listed in Takeuchi's, are implicitly mentioned in the term "refractory metal". (Partial of this argument is also responded as in the office action mailed 04/03/2007, section 5.)

Appellant also rehashes the argument on the percentage of nickel in the alloy used to form the silicide layer. Another argument from the Advisory partially response to this:

In response to applicant's statement that "the Maex reference cannot cure the above deficiencies" ... The extracts in the rejection wherein the Maex reference discloses

The first layer structure can also include a cobalt-nickel alloy with the nickel content varying from 0 to 100%; ... Also, other metals such as Pt or Pd can be chosen as elements that are present in the first layer structure.... or the elements Pt and Pd can be added in minor amounts to the first layer structure. Also, other elements such as Au, Ir, Os, Rh, Ti, Ta, W, Mo, Cr, C, and Ge can be part of the first layer structure.

This passage means that with 100% nickel, the material is not all cobalt; with less than 100% nickel and the added material, in minor amounts, of one of Ta, Pd, Pt, e.g., the combination satisfies the requirement of all claims.

In response to Appellant's argument that there is no suggestion to combine the references, the examiner recognizes that obviousness can only be established by combining or modifying the teachings of the prior art to produce the claimed invention where there is some teaching, suggestion, or motivation to do so found either in the references themselves or in the knowledge generally available to one of ordinary skill in the art. See *In re Fine*, 837 F.2d 1071, 5 USPQ2d 1596 (Fed. Cir. 1988) and *In re Jones*, 958 F.2d 347, 21 USPQ2d 1941 (Fed. Cir. 1992). In this case, the examiner, throughout the rejection, strictly follows the four factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a).

Appellant's Arguments in 7.B. (i) are the same as in 7.A. (i) and are response the same.

(11) Related Proceeding(s) Appendix

No decision rendered by a court or the Board is identified by the examiner in the Related Appeals and Interferences section of this examiner's answer.

For the above reasons, it is believed that the rejections should be sustained.

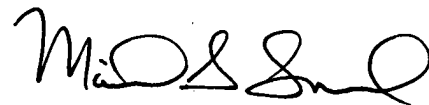
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